

## Collision Theory Worksheet Answers

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## **collision theory worksheet answers**

of Reaction Mindset Learn 1 1 Rate of reaction 1 2 Collision theory Task Answers Question 1 1  
1 Rate of reaction is the change in concentration of products per unit time 1 2 Collision theory  
explains the conditions for the reacting particles to collide together in a chemical A Guide to  
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Observe a chemical reaction with and without a catalyst Determine the effects of concentration  
temperature surface area and catalysts on reaction rates Reactant and product concentrations  
through time are recorded and the speed of the simulation can be adjusted by the user  
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without reacting If we could watch every collision we might see that only 1 collision in 10 or in  
100 had both molecules properly oriented for reaction A second factor is the energy of the two  
molecules In the simplest theories this is expressed only as the relative speed of the two  
molecules upon collision Chapter 16 Reaction Rates Middlesex County Vocational 560  
Chapter 16 • Reaction Rates Section 116 16 1 A Model for Reaction Rates MAIN Idea  
Collision theory is the key to understanding why some reactions are faster than others Real  
World Reading Link Which is faster walking to school or riding in a bus

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Author : Sven Strauss

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